

Comparison of *Nigella sativa* Seeds Extract and Respective ZnO NPs for Antimicrobial, Antioxidant and Cytotoxic Potential

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Abstract. In the present work, ZnO NPs were synthesized by using zinc acetate salt solution and seeds extract of *Nigella sativa* L. in a ratio of 9:1, for a green chemistry approach being nontoxic and ecofriendly in nature. Colour change confirmed the formation of ZnO NPs. The synthesized nano structures characterized by UV-Vis spectrophotometer, SEM, EDX, FTIR and XRD. To evaluate the therapeutic efficacy of ZnO NPs various bioassays were performed such as antifungal, antibacterial, antioxidant and brine shrimp cytotoxicity. UV-Vis spectrophotometer generated the absorption peak at 370 nm for ZnO NPs. SEM results indicated the average size of ZnO NPs 35 nm with a hexagonal and crystalline structure which was confirmed through XRD. The active functional group of amines, methyl and OH⁻ were analyzed through FTIR. Zinc is the major constituent in the sample confirmed by the EDX spectrum. Biologically synthesized nanoparticles of *Nigella sativa* seeds extract showed significant antibacterial activity at 100 ppm, 80% antifungal activity at 25 ppm as compared to plant extract (15%). ZnO NPs also showed significant antioxidant potential (70%) at 100 ppm as compared to plant extract (51.72%) with EC₅₀ 65.06 ppm for plant extract and 53.46 ppm for ZnO NPs. Significant cytotoxicity potential also observed against the brine shrimps with IC₅₀ of ZnO NPs 147.4 ppm as compared to plant extract with IC₅₀ 258.9 ppm. This is the first report stating the synthesis and antimicrobial, antioxidant and cytotoxic potential of ZnO NPs of *Nigella sativa* seed extract. These ZnO NPs can be further explored in various biomedical applications at the commercial level.

Keywords: antimicrobial, antioxidant, cytotoxic, *Nigella sativa*, zinc oxide nanoparticles

Introduction

Nanotechnology is the most advanced and emerging field in the present era. It encircles the application and production of nanoscale material in biological, chemical and physical systems (Murphy *et al.*, 2008). It is also helpful in comparing cellular and molecular biology, information technology and semiconductor technology. Due to its potential, it has a positive impact on the life of humans and the treatment of multiple diseases (Ghosheh *et al.*, 1999). Researchers in nano-technology have focused on the manufacturing of technically designed nano-materials, that can actively participate at the sub-cellular level of the body with high specificity as nano-medicines (Ahmad *et al.*, 2008). The Zinc oxide nano-particles (ZnO NPs) have received much attention due to their astonishing properties and characteristics, these nanoparticles showed the ability of cytotoxicity in rapidly dividing cancer cells. They possess mechanical and thermal stability in chemical reactions and biological

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species (Zhang *et al.*, 2011). This property of ZnO NPs makes them more attractive and effective in modern laser medical technology, electronics and opto-electronics.

Production of ZnO NPs by the method of green synthesis makes them bio-compatible, low toxic, and ecofriendly with increased applications in bio-medicine (Bacaksiz *et al.*, 2008; Wang *et al.*, 2005). The plant appears to be the leading and suitable platform for the synthesis of large-scale bio-synthesized nanoparticles (Albanese *et al.*, 2012). Physical and chemical approaches are less used for the synthesis of ZnO NPs because these methods create a lot of problems with cost, bio-compatibility, particle size, size distribution and yields (Patra and Baek, 2014). Medicinal plants have played an effective role for thousands of years in human society. They are the only main source of natural pure phyto constituents. These phyto constituents are most commonly used in the treatment of various disorders, preventing epidemic diseases and also as food preservatives and flavour condiments (Silva and Fernandes Júnior, 2010).

Nigella sativa (*N. sativa*) is one of the miracle plants that has rich historical and religious values. It is an annual herbaceous medicinal plant that belongs to the family Ranunculaceae, commonly known as 'black cummin' (Imran *et al.*, 2022). In ancient times, *N. sativa* was most intensively used in different medical systems such as Ayurvedic, Chinese, Unani and Arabic for the curing of several disorders (Randhawa and Alghamdi, 2011). The extracts of *N. sativa* seeds actively played a therapeutic role in retarding the inflammation process and antioxidants activation, stopping the carcinogenic process, restraining the cough in patients, treating diarrhoea, abdominal pain, flatulence and polio (Al-Sheddi *et al.*, 2014). The seed of *N. sativa* also contains proteins, saponin, alkaloids and essential oils (Khan *et al.*, 2011). It also played an effective role in appetite stimulation, as a liver tonic and is helpful in the digestion of food, enhancing the milk production quantity in nursing mothers and to fight against the pathogens and supporting the immune system (Abel-Salam, 2012; Assayed, 2010; Abdel-Sater, 2009). The therapeutic properties of these plants are based on the main active phyto-constituents, thymoquinone (TQ) which is the main constituent of *N. sativa* seeds oil (Abdel-Zaher *et al.*, 2011, Boskabady *et al.*, 2010).

Materials and Methods

Synthesis of ZnO NPs. *N. sativa* seeds (20 g) were boiled in 100 mL distilled water in a 500 mL beaker for 30 min. The Whatman No.1 filter paper was used to eliminate the debris. Moreover, zinc acetate salt solution 0.25 M was prepared by adding 22.92 g zinc acetate salt in 500 mL of distilled water in the laboratory. The biosynthesis of ZnO NPs was done by mixing zinc acetate salt solution and seeds extracts of *N. sativa* in a ratio of 9:1. 5 mL seed extract was dissolved in 45 mL of zinc acetate salt manually on magnetic stirrers for 25 mins (Alaghemand *et al.*, 2018). The reaction mixture was subjected to room temperature until its colour changed under sunlight. The colour change indicated the formation of the nanoparticles. All experiments were carried out in triplicate. The pellet of ZnO NPs was dried in a hot air oven at 60 °C temperature for up to 48 h (Gabriela *et al.*, 2017). The ZnO NPs were obtained in the powdered form which was collected and preserved in the Eppendorf Tubes (Hashemi *et al.*, 2016).

Characterization of ZnO NPs. UV-Vis spectrophotometer was used to recognize the optical properties of

zinc oxide nanoparticles. The spectra were observed at 320-440 nm (Swamy *et al.*, 2015; Gilani *et al.*, 2004). The morphology of the synthesized ZnO NPs was identified by using Scanning Electron Microscope (JEOL-JSM-6490LATM) at the voltage of 20Kv with a frequency of 2838 cps (max) (Wang *et al.*, 2005). Energy-dispersive X-ray spectroscopy was carried out for synthesized ZnO NPs to determine the elemental composition and their proportion in a sample that was coupled with Scanning Electron Microscope (SEM). Analysis of surface chemistry of ZnO NPs based on FTIR was performed. In the FTIR spectrometer, the solution was dried at the degree of 75 °C and characterization was done at the ranges of 4000-400 cm⁻¹ by using the KBr pellet method (Alamdari *et al.*, 2020). X-ray was bombarded on the targeted sample to analyze the structure of ZnO NPs. An X-ray diffractometer was used to generate an X-ray spectrum (Whitfield and Mitchell, 2004). The energy of the beam ranges in between 10-20 KeV which causes the X-rays to be emitted from samples. The electron beam moves across the samples and images are obtained for the synthesized ZnO NPs (Stevenson *et al.*, 2005).

Antibacterial assay. The method used to analyze the antibacterial activity is a disc diffusion method described by (Ruparelia *et al.*, 2008). Biologically synthesized zinc oxide nanoparticles used to determine the antibacterial activity against the gram-negative strains AT 10, *Enterobacter aerogenes*, *Salmonella setubal* and gram-positive strains *Bacillus subtilis*, *Salmonella aureus*, *Micrococcus letus*. The pellet (25 mg) of synthesized ZnO NPs was dissolved in the distilled water (25 mL) and the final concentration of 1000 ppm was made. During the antimicrobial assay, 10, 20, 30, 40, 50 and 100 ppm were used. Each Petri plates contain 8 discs, 6 discs contain the different concentration ZnO NPs samples in triplicates. Distilled water was used as negative control and streptomycin as a positive control. Discs were arranged on the solidified broth plates in a sequence. The Petri dishes were sealed and incubated for 24 h at 37 °C in a dark environment. After the incubation period of 24 h, the region of inhibition was observed in every disc and the zone of inhibition around every disc was measured (Huys *et al.*, 2002).

Antifungal assay. The tube dilution method was used to determine the antifungal activity of biologically synthesized ZnO NPs (Singh and Nanda, 2013). There are the following strains that were used for the fungal activity *Aspergillus flavis*, *Aspergillus fumigates*,

Aspergillus niger, *Mucor ans Fusarium solani*, 4 mL of sabouraud dextrose agar was poured into each test tube to make a slant and cotton swabs were used to cover these test tubes. After that 100 μ L of the sample (25 ppm final working concentration) was added to these test tubes. Slant was made to the 10 cm mark at room temperature inoculation loop was used for the introduction of fungal strains. The cotton plugs were used to cover. The whole experiment was performed in triplicate for each sample. Antifungal drug terbinafine was utilized as positive control and distilled water was taken as a negative control. The next step was the incubation of these tubes at 37 °C for 4 days. Negative control was taken as a reference for the determination of the fungal growth in a linear position. Reading was documented measuring the fungal growth in slanting position. The subsequent formula was utilized to calculate the proportion of fungal growth inhibition.

$$\text{Inhibition} = \frac{\text{Linear growth in negative control} - \text{linear growth in samples}}{\text{Linear growth in the negative control}} \times 100$$

Antioxidant assay. DPPH method (2,2-diphenyl-1-picrylhydrazyl-hydrate) was used to determine the antioxidant activity of biologically prepared ZnO NPs and extraction of *N. sativa* seeds that was descriptively mentioned earlier (Gyamfi *et al.*, 1999). 200 μ L serial solution of synthesized nanoparticles was added along with 2.8 mL of DPPH reagent. Ethanol was taken as a negative control and ascorbic acid was used as a positive control. Glass vials containing solutions were put in a dark environment for 50 min to determine the antioxidant property of ZnONPs. Ethanol blank was used as reference and absorbance of the sample was measured at 517 nm. The subsequent formula was applied to calculate the scavenging percentage of free radicals.

$$\% \text{ Scavenging} = \frac{\text{Control absorbance} - \text{nanoparticle samples absorbance}}{\text{Absorbance of control}} \times 100$$

Brine shrimp lethality assay. The cytotoxic potential of prepared ZnONPs was determined by using brine shrimps lethality assay by following the method reported earlier (Bibi *et al.*, 2011). In the first step, nanoparticles were added in glass vials at different concentrations of 25, 50 and 100 ppm with sea salt solution to make the final volume 5 mL. Distilled water was taken as a negative control. After 24 h, most of the shrimps were hatched from the eggs and young tiny shrimps were

floating on the surface of sea water. They were shifted in each vials 15 in numbers. After the completion of this process, the glass vials were put at room temperature at 25 °C. After 24 h, a pasture pipette (3 \times magnifying glass) was used to count alive shrimps. Percentage mortality was calculated by the given formula, the whole procedure was carried out three times.

$$\% \text{ Mortality} = \frac{\text{Number of alive shrimps in negative control} - \text{number of alive shrimps test}}{\text{Number of alive shrimps in the negative control}} \times 100$$

Results and Discussion

Synthesis and characterization. Change in the colour of the solution from colourless to light brown or yellow is the confirmatory sign of ZnONPs formation. This changed colour was observed after the 10 min when the mixing of zinc acetate salts solution was done with *N. sativa* seeds extracts (Gabriela *et al.*, 2017). The extract of the plant reduced the zinc acetate salt into ZnO NPs. Furthermore, the ZnO NPs synthesis was tested for optical properties by the UV-Vis spectrophotometer. The wavelength of the UV-Vis spectrophotometer was set in between 300-500 nm. The spectra observed at 370 nm indicated ZnO NPs synthesis (Fig. 1) (Kumar *et al.*, 2004). The size and morphology of synthesized ZnO NPs were observed by the SME analysis. The average size of ZnO NPs was observed to be 35 nm (Fig. 2) (Mohan and Renjanadevi, 2016). EDX (Energy dispersive X-ray spectroscopy) in operation with SEM was used to conclude the elemental composition, arrangements and chemical composition of biological synthesized ZnO NPs. The spectrum of EDX also confirmed the Zinc and oxygen as the main constituents without any contamination. The corresponding peak except for the oxygen and zinc *i.e.* carbon may be because carbon coated gird capping agents were utilized in scanning electron microscope measurements shows in (Fig. 3) (Tao *et al.*, 2006). The crystalline alignment phase composition and phase identification of biologically synthesized ZnO NPs from *N. sativa* seed extract was confirmed by the XRD pattern. XRD pattern of the ZnO sample was observed by using index POWDER-X software as well as matched with standard JCPDS, 36-1451 data. The result showed that synthesized ZnO NPs have hexagonal wurtzite structure with crystalline nature with identification peaks having lattice parameter $a=3.252(3)$ (Å), $c=5.208(6)$ (Å). No other impurity/extra peak was detected which indicates

that obtained product is of high purity. The crystalline measurement was done by using Scherrer equation $D_c = 0.9 \lambda / \beta \cos \theta$ (Fig. 4) (Alaghemand *et al.*, 2018; Kavyashree *et al.*, 2015). *N. sativa* seeds were actively involved in the synthesis of ZnO NPs. In the modification of Zn-O bonding, the FTIR technique has been implemented for ZnO NPs. At room temperature, the KBR method was used to perform the FTIR measurement in the wave range of $4000\text{--}400\text{ cm}^{-1}$ (Al-Shabib *et al.*, 2016). The FTIR spectrum showed the main band near 3225 cm^{-1} which represents the O-H (covalent bond) mode and FTIR represents the methyl rock mode at the band of 1384 cm^{-1} . Another mode N-H with a hydrogen

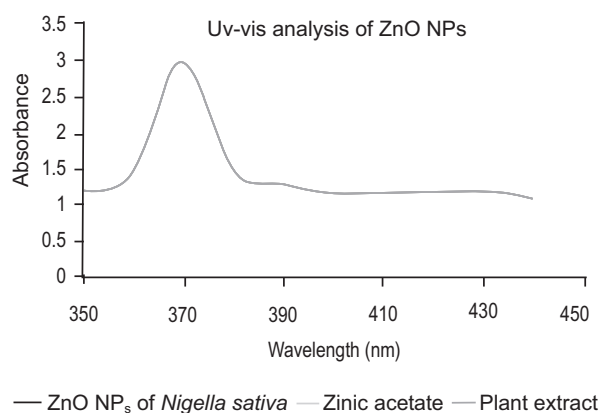


Fig. 1. UV-Vis spectroscopy. The spectra show the absorption peaks of ZnO NPs in the range of 350–450 nm.

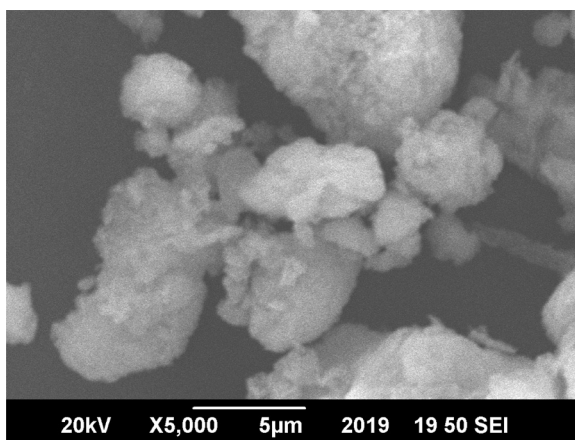


Fig. 2. Scanning electron microscopy (SEM) micrographs of ZnO NPs. The images are presenting the shape and size of the synthesized nanoparticles. Scale bar = 5 μm .

bond was observed at the band of 690 cm^{-1} (Fig. 5) reported by (Manju *et al.*, 2016). This can be identified as the proteins of this plant are involved in ZnO NPs synthesis.

Antibacterial assay. The antibacterial activity of biologically synthesized ZnO NPs (10, 20, 30, 40, 50 and 100 ppm) was performed against six strains

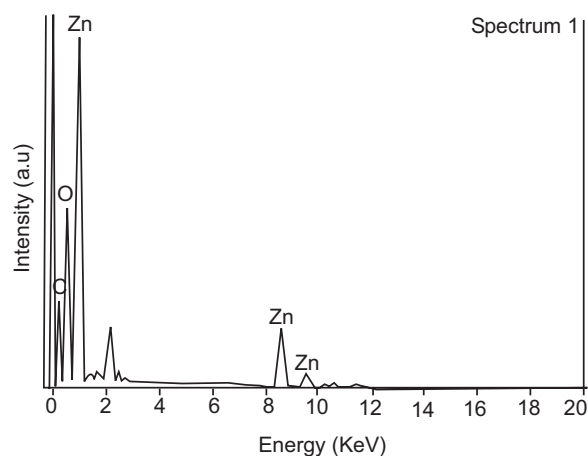


Fig. 3. Energy-dispersive X-ray spectroscopy, the EDS analysis of synthesized ZnO NPs and the spectrum shows Zn and O as the major component of nanoparticles.

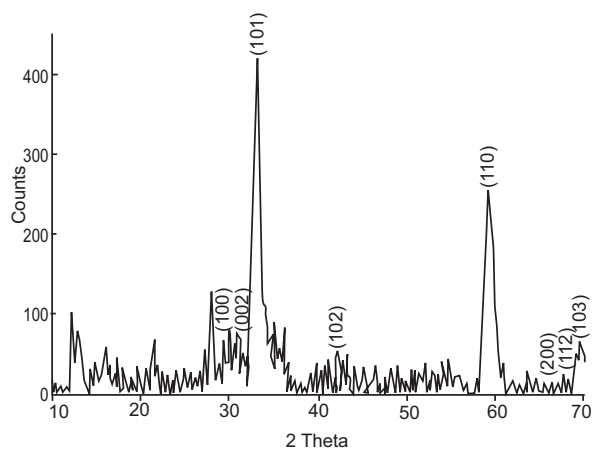


Fig. 4. X-ray diffraction (XRD) analysis of synthesized ZnO NPs. The spectrum shows hexagonal wurtzite structure with crystalline nature of ZnO NPs. The intensity on the vertical axis is measured in counts per second (CPS), and the diffraction angle (2θ) measured is taken along the horizontal axis.

Table 1. Antibacterial activity of synthesized ZnO NPs and plant seeds extracts (*Nigella sativa*)

ZnO NPs	Zone of inhibition (cm) ± S. E.											
	Gram positive strains						Gram negative strains					
	<i>M. luteus</i>		<i>S. aureus</i>		<i>B. subtilis</i>		<i>A. tumefaciens</i>		<i>S. setubal</i>		<i>E. aerogenes</i>	
Con (ppm)	ZnO NPs	P. extract	ZnO NPs	P. extract	ZnO NPs	P. extract	ZnO NPs	P. extract	ZnO NPs	P. extract	ZnO NPs	P. extract
10	1.2±0.1	-	-	-	0.7±0.1	-	-	-	1±0.15	-	0.8±0.1	-
20	1.4±0.1	-	-	-	1.9±0.5	-	-	-	1.2±0.1	-	1.2±0.1	-
30	1.6±0.15	-	1.5±0.1	-	1±0.1	-	-	-	1.5±0.3	-	1.7±0.1	-
40	1.9±0.1	-	1.5±0.1	-	1.5±0.1	-	-	-	1.6±0.1	-	2±2.0	-
50	2±0.1	-	1.8±0.1	-	2.5±0.1	-	-	-	1.8±0.1	-	2.5±0.3	-
100	2.3±0.1	15±0.1	2±0.1	11±0.1	3±0.05	12±0.1	1±0.15	-	2±0.1	1±0.1	2.5±0.1	11±0.1
-ve control	0	0	0	0	0	0	0	0	0	0	0	0
+ve control	3.1	3	2.7	2.7	3.5	3.2	1	1	3	3	3	3
100 ppm												

ZnO NPs = zinc oxide nanoparticles; Conc = concentration; S.E = standard error; *M. luteus* = *Micrococcus luteus*; *S. aureus* = *Staphylococcus aureus*; *B. subtilis* = *Bacillus subtilis*; *A. tumefaciens* = *Agrobacterium tumefaciens*; *S. Setubal* = *Salmonella setubal*; *E. aerogenes* = *Enterobacter aerogenes*.

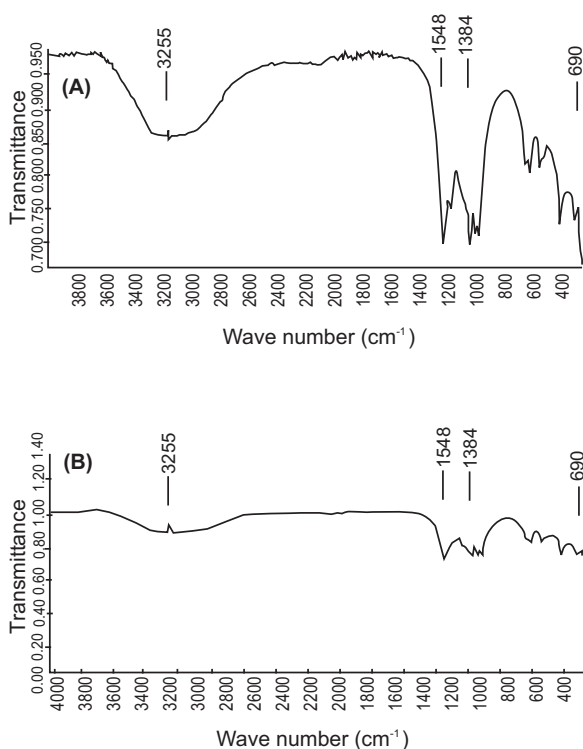


Fig. 5. FTIR analysis. FTIR analysis of plant extract (A) and ZnO NPs (B). The FTIR spectrum showed the main band near 3225 cm⁻¹ which represents the O-H (covalent bond) mode, and also represents the methyl rock mode at the band of 1384cm⁻¹. Another mode N-H (primary and secondary amines only) with hydrogen bond was observed at the band of 690 cm⁻¹.

i.e. three-gram negatives (*A. tumefaciens*, *S. setubal*, *E. aerogenes*) and three gram-positive strains (*M. luteus*, *S. aureus*, *B. subtilis*). Results are mentioned in (Table 1). It was distinct that by increasing the concentration of ZnO NPs, a progressive increase in the inhibition of bacterial growth was observed (Wang *et al.*, 2012). ZnO NPs were also tested to determine their minimum inhibitory concentration (MIC). The highest zone of inhibition was observed at 100 ppm in both strains gram-positive (*M. luteus*, *S. aureus*, *B. subtilis*) and gram-negative (*A. tumefaciens*, *S. Setubal*, *E. aerogenes*). Very minute inhibition activity was observed at 10 ppm of ZnO NPs in (gram-positive and gram-negative strains) (Emami-Karvani and Chehrazi, 2011). Moreover, ZnO NPs were also found more effective as compared to plant extract. It happens when the biologically synthesized ZnO NPs element interacts with the microbial cell surface (Djurišić *et al.*, 2015). Teichoic acid in the peptidoglycan layer of the gram-positive and lipoteichoic acid in gram-negative bacteria, both due to rich polyphosphate anions played as effective mediators for the entry of ZnO NPs (Xu *et al.*, 2013; Hajipour *et al.*, 2012). It was reported that ZnO nanoparticles exhibit antibacterial properties because of increased surface area to volume ratio and increased reactivity possibly by the mechanism of generation of reactive oxygen species. Oxidative stress produces free radical OH⁻ group and H₂O₂ that damage the DNA, cell membrane integrity as well as metabolic and enzymatic functions of bacteria leading to ultimately cell death (Sirelkhatim *et al.*, 2015).

Antifungal assay. The experiment was conducted against the different strains of fungi which are responsible for various diseases by using biologically prepared ZnO NPs and the *N. sativa* seeds extract. It was observed that ZnO NPs at 25 ppm showed maximum inhibitory effect against the *Fusarium solani* 80% and minimum inhibitions were observed against the *Mucor* species 20%. The ZnO NPs also showed moderate inhibition against the remaining three strains *Niger*, *Fumigatus*, *flavis* which are 40, 40 and 50%, respectively. The results are mentioned in (Table 2). Distilled water was used as negative control and terbinafine for the positive control. The extracts of *N. sativa* seeds were also used in comparison against the ZnO NPs. Plant extract of *N. sativa* seeds only showed a significant effect against three strains of fungi except for the two strains *Mucor* and *Niger* species. In the case of seeds extract, the maximum antifungal activity was observed against *Solnia* 25%, *Flavis* 20% and *Fumigates* 15% (Table 2). *N. sativa* seeds extracts did not show any antifungal activity against *Mucor* 0% and *Niger* 0% (Kairyte *et al.*, 2013).

After the comparison, it was proved that the synthesized ZnO NPs due to small size large surface to volume ratio and attachment of active secondary metabolites thymol, thymohydroquinone, saponins and alkaloids showed maximum antifungal effects as compared to just plant seeds extracts (Sharma *et al.*, 2010).

Antioxidant assay (DPPH). To evaluate the antioxidant ability of synthesized ZnO NPs, the experiment was

done by the DPPH (2,2 diphenyl-1-1 picrylhydrazyl hydrate) assay. The synthesized ZnO NPs showed antioxidant properties by donating an electron of oxygen toward the hydrogen. DPPH is one of the stable free radicals and can accept the electron from synthesized ZnO NPs and become stable diamagnetic molecules (Das *et al.*, 2013). In this study, a comparison of synthesized ZnO NPs and seed extract of *N. sativa* was done. ZnO NPs revealed more antioxidant-free radical scavenging activity as compared to seeds extract. At 100 ppm concentration, the hexagonal ZnO NPs exhibited antioxidant scavenging free radicals up to 70.11%, whereas the extract of subjected seeds *N. sativa* plant at 100 ppm concentration showed 51.72% free radical scavenging activity. At 50 ppm the ZnO NPs exhibited 58.6% free radical scavenging activity and extract of *N. sativa* seeds showed 33.3% free radical scavenging activity (Hashemi *et al.*, 2016). At the very lowest concentration, 25 ppm ZnO NPs pretend 49.9% antioxidant free radical scavenging activity and extract of *N. sativa* seeds showed 19.9% free radicals scavenging ability (Lee *et al.*, 2010). The results were also found quite significant statistically ($P < 0.0001$) (Table 3). EC50 53.46 ppm for zinc oxide nanoparticles was observed. However, plant extracts had a greater EC50 65.06 ppm which proved that ZnO NPs are more promising antioxidants as compared to plant extract (Fig. 6). To confirm the stability of the DPPH solution the solution was kept for 2 h without any disturbance. There was no change of colour of the solution that indicated the maximum stability of DPPH during the experiment (Das *et al.*,

Table 2. Percentage inhibition of ZnO NPs and plant seeds extracts (*Nigella sativa*) against different fungal species

Samples	Percentage inhibition against fungal species (%)				
	<i>Mucor.sp</i>	<i>F. Solani</i>	<i>A. Fumigatus</i>	<i>A. Flavis</i>	<i>A. Niger</i>
ZnO NPs	20	80	40	50	40
Plant extract	-	25	15	20	-
Distilled water (-ve Cont)	-	-	-	-	-
Terbinafine (+ve Cont)	100	100	100	100	100

ZnONPsz = zinc oxide nanoparticles; *Mucor. sp* = *Mucor* species; *F. Solani* = *Fusarium solani*; *A. fumigatus* = *Aspergillus fumigatus*; *A. Flavis* = *Aspergillus flavis*; *A. Niger* = *Aspergillus niger*.

Table 3. Analysis of variance for factors affecting the free radical scavenging activity samples

Source of variation	Df	Sum of squares	Mean square	F-Value	P-value	Significant
Interaction	4	846.5	211.6	27.25	<0.0001	Yes
ZnO Nano particles and seeds extract	2	19110	9553	1230	<0.0001	Yes
Concentration	2	1653	826.4	106.4	<0.0001	Yes
Residual	18	139.8	7.767	-	-	-

2013). The absorption intensity was observed at 517 nm. When the ZnO NPs were added to the DPPH solution the absorption peak intensity was gradually decreased at 517 nm and the colour of the solution gradually changed from deep violet to pale yellow. The peak intensity at 517 nm of DPPH is inversely proportional to the amount of ZnO NPs. Antioxidant activity of ZnO NPs was based on the transfer of electrons located on oxygen atom toward hydrogen atom of DPPH due to this, the transition peak of DPPH at 517 nm decreased (Lewicka and Colvin, 2013). Biologically synthesized ZnO NPs and *N. sativa* seeds extract showed a significant impact as ($P < 0.01$) against the free radical scavenging activity shown in (Fig. 6) (Table 3). This antioxidant activity of ZnO NPs and seed extract might be because of the bioactive compounds of *N. sativa* which were also utilized in the synthesis of ZnO NPs.

Cytotoxic assay. To evaluate the toxicity of ZnO NPs, the cytotoxic assay was performed on the brine shrimps. The different concentrations were used for ZnO NPs and plant seeds extract *i.e.* 100, 50 and 25 ppm, which showed significant toxic effects (Hameed *et al.*, 2019). At the start of the cytotoxic assay to check the mortality of brine shrimps against the plant extract and ZnO NPs maximum concentration was taken. At the concentration of 100 ppm, ZnO NPs illustrated the 46% mortality of brine shrimp and *N. sativa* seeds extract showed 38%. At 50 ppm concentration, ZnO NPs confirmed the 33.3% mortality and plant seeds extract showed 26.6% mortality. At the concentration of 25 ppm, ZnO NPs proved 20% mortality and plant seeds showed 6.6% mortality. After comparison between the plant seeds extracts of *N. sativa* and synthesized ZnO NPs it was observed that ZnO NPs showed more mortality rate at 100 ppm 46% as compared to plant seeds extract at 38% (Fig. 7). It was also found that higher concentration corresponds to a higher mortality rate (Fig. 7). The result was proved significant statistically ($P < 0.001$) (Table 4). After results analysis, it was observed that IC_{50} value ZnO NPs 147.4 ppm is less as compared to seeds extract of *N. sativa* IC_{50} value 258.9 ppm. ZnO NPs showed more significant results and hence can be used on cancer lines. It was suggested that a higher concentration of ZnO NPs directly related to the mortality rate of Brine shrimps. In an early study, ZnO NPs were reported to have maximum cytotoxicity due to several factors such as dose, time, small size large surface-to-volume ratio of ZnO NPs (Najim *et al.*, 2014). Moreover,

it was observed that biologically synthesized ZnO NPs from *N. sativa* seeds extract showed maximum anti-cancerous activity against the panel of human cancer cell lines (Nazrul Islam *et al.*, 2004). It has also been observed ZnO NPs generated oxidative stress against the MCF-7 breast cancer cell lines and acted as an anti-cancerous agent (Awad, 2005). Moreover, ZnO NPs induced toxicity against the retinal ganglion cells. It was suggested that ZnO NPs inside the retinal ganglion cell lose the membrane potential of mitochondria and enhanced the over production of reactive oxygen species

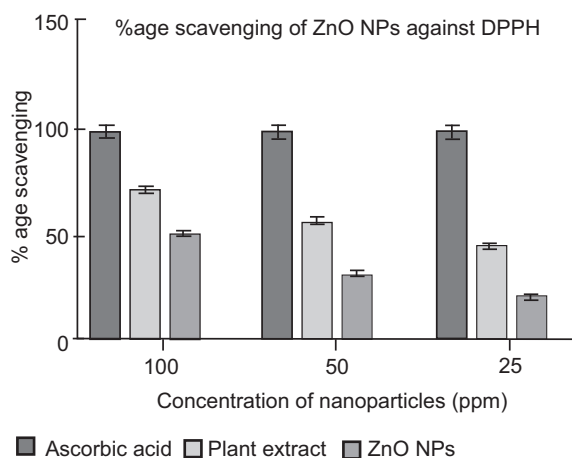


Fig. 6. Antioxidant activity of ZnO NPs and plant extract. Free radical scavenging or antioxidant activity determined by 2,2-diphenyl-1-picrylhydrazyl (DPPH) assay.

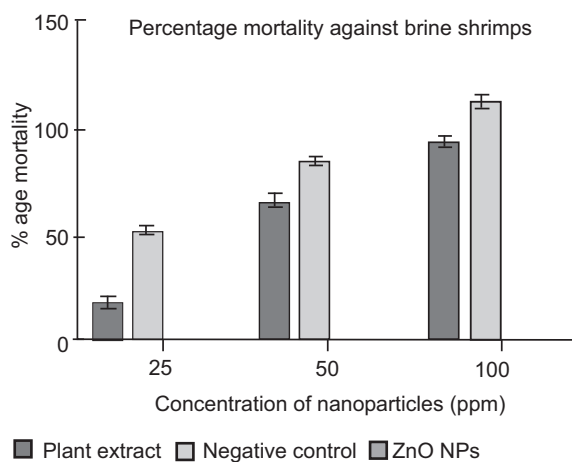


Fig. 7. Cytotoxicity of ZnO NPs and plant extract. cytotoxicity of ZnO particles against brine shrimps.

Table 4. Analysis of variance for factor affecting the mortality of brine shrimps

Source of variation	Df	Sum of squares	Mean square	F-Value	P-Value	Significant
Interaction	4	1154	288.4	7.819	<0.0001	Yes
ZnO Nano particles and seeds extract	2	21790	10890	295.3	<0.0001	Yes
Concentration	2	2339	1169	31.70	<0.0001	Yes
Residual	18	664.0	36.89	—	—	—

(ROS) and also enhanced the caspase-12 in RGC-5 cells. This oxidative stress and over disturbance triggered the endoplasmic reticulum to promote the RGC-5 damage and finally induced apoptosis or necrosis (Guo *et al.*, 2013).

Conclusion

N. sativais a medicinal plant with a common name (black cumin) and seeds extract was used for the biological synthesis of ZnO NPs. It was analyzed that biologically active compounds of plants played a vital role in the synthesis process and act as reducing and capping agents. Synthesized ZnO NPs were found active members in antimicrobial, antifungal, antioxidant and cytotoxic processes and can be explored further as (bioimaging, antibacterial, antifungal, anticancerous drugs). After the evaluation of synthesized ZnO NPs through Uv-Vis, SEM, XRD, FTIR, cytotoxic and antioxidant assay, antibacterial and antifungal assays it was proved that bio-activities of ZnO NPs were enhanced as compared to plant extract with less IC₅₀ values. So, these stable metallic nanoparticles can also be used in nano-medicines, controlled drug treatments in cancer, and appropriate targeted drug delivery in the future.

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Conflict of Interest. The authors declare they have no conflict of interest.

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